

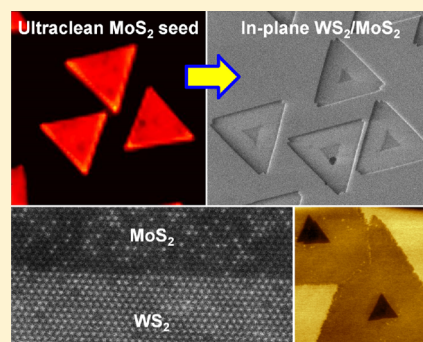
Seed Crystal Homogeneity Controls Lateral and Vertical Heteroepitaxy of Monolayer MoS₂ and WS₂

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S Supporting Information

ABSTRACT: Heteroepitaxy between transition-metal dichalcogenide (TMDC) monolayers can fabricate atomically thin semiconductor heterojunctions without interfacial contamination, which are essential for next-generation electronics and optoelectronics. Here we report a controllable two-step chemical vapor deposition (CVD) process for lateral and vertical heteroepitaxy between monolayer WS₂ and MoS₂ on a *c*-cut sapphire substrate. Lateral and vertical heteroepitaxy can be selectively achieved by carefully controlling the growth of MoS₂ monolayers that are used as two-dimensional (2D) seed crystals. Using hydrogen as a carrier gas, we synthesize ultraclean MoS₂ monolayers, which enable lateral heteroepitaxial growth of monolayer WS₂ from the MoS₂ edges to create atomically coherent and sharp in-plane WS₂/MoS₂ heterojunctions. When no hydrogen is used, we obtain MoS₂ monolayers decorated with small particles along the edges, inducing vertical heteroepitaxial growth of monolayer WS₂ on top of the MoS₂ to form vertical WS₂/MoS₂ heterojunctions. Our lateral and vertical atomic layer heteroepitaxy steered by seed defect engineering opens up a new route toward atomically controlled fabrication of 2D heterojunction architectures.



INTRODUCTION

Two-dimensional (2D) materials have emergent electronic and optical properties that are distinct from those of their bulk counterparts, as exemplified by graphene and its parent bulk material, graphite.^{1–3} Monolayer transition-metal dichalcogenides (TMDCs) are a class of 2D materials that have direct band gaps and broken inversion symmetry, which are absent in the bulk.^{4–8} These two combined properties have stimulated interest in using these materials for photodetectors, field effect transistors, and developing new devices to measure the physics of valley polarization.^{9–15} Heterostructures between 2D materials exhibit junctions possessing new properties that are unobtainable from single-component 2D materials. In particular, heterojunctions composed of TMDC monolayers can serve as building blocks for next-generation optoelectronic devices, such as atomically thin solar cells and light-emitting diodes, due to their strong light–matter interactions.^{14,16} The first heterojunctions between TMDC monolayers were created by mechanical exfoliation and vertical stacking.^{17,18} Stacking WS₂/MoS₂ and stacking WSe₂/MoS₂ create type II heterostructures whose electronic properties depend sensitively on the twist angle between the two layers.^{17–20} Recently, it was reported that both vertical and in-plane heterostructures of monolayer WS₂/MoS₂ can be synthesized using chemical vapor deposition (CVD) in a single step with MoO₃, W, tellurium, and sulfur precursors.²¹

Heteroepitaxy, the oriented growth of one crystalline material on another, can realize the scalable fabrication of atomically perfect heterojunctions without interfacial contamination.²² In addition, heteroepitaxy can create in-plane TMDC

heterojunctions that cannot be obtained by mechanical transfer methods. These in-plane junctions have intriguing optoelectronic properties, including a linearly abrupt p–n junction,²¹ similar to junctions found in commercial field effect transistors. The previously reported methods for creating these heterojunctions utilized a single synthetic step,^{21,23} which limits ultimate control over important features, such as particle size, shape, location, and junction width. A two-step process is highly desirable, because it could realize patterned 2D heterostructures and also achieve independent growth control of each component material. The advantages of controlled multistep growth have been previously reported for the synthesis of heterostructures composed of graphene and hexagonal boron nitride.^{24–28}

Fabricating in-plane 2D heterostructures using two-step heteroepitaxy is challenging due to difficulties in controlling defects and contamination of the 2D seed crystals. Defects or particles persisting after the growth of the first material will serve as undesirable nucleation sites for depositing the second material.²⁹ Consequently, a two-step process to fabricate in-plane heterostructures composed of monolayer WS₂/MoS₂ requires extra care in preparing extremely clean surfaces and edges of the 2D seed crystals. Previous reports of CVD-grown TMDC monolayers, especially MoS₂, have shown that the surface and edges of individual monolayers are often contaminated by small particles.^{30–37}

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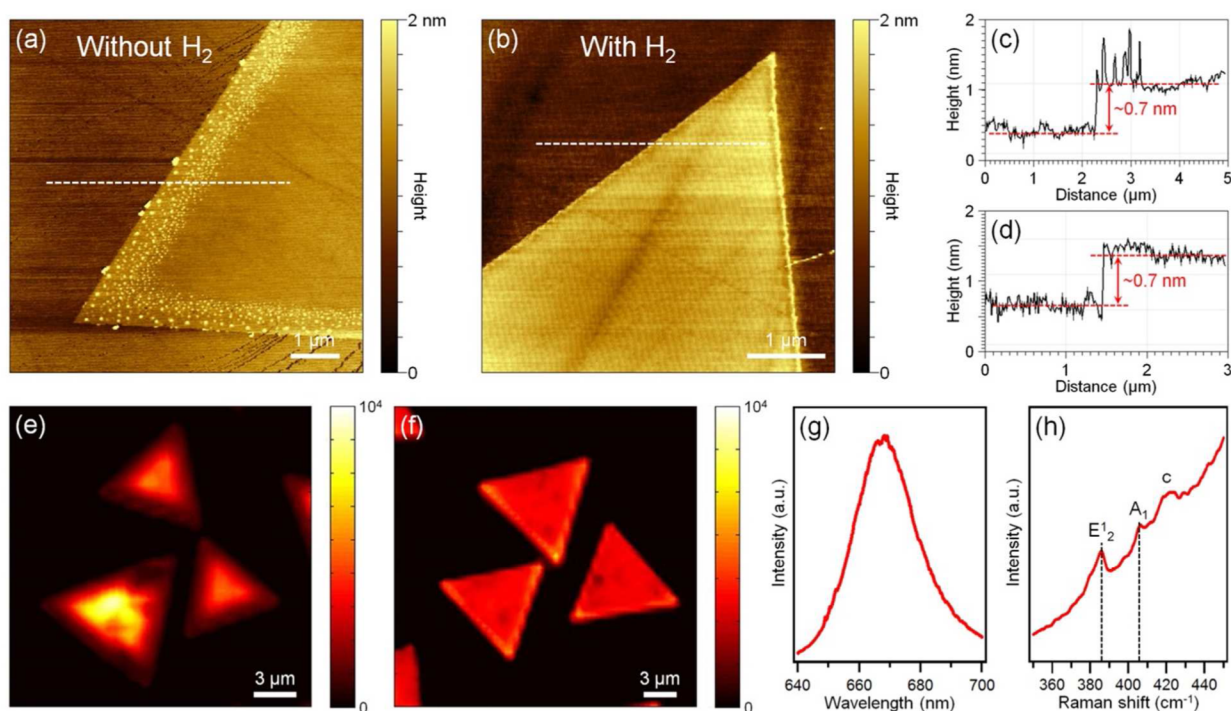


Figure 1. Monolayer MoS₂ crystals synthesized on *c*-cut sapphire without and with hydrogen gas. AFM height images of monolayer MoS₂ grown (a) without and (b) with hydrogen, respectively. (c, d) Height line profiles along the dotted white lines in panels a and b, respectively. PL intensity maps of monolayer MoS₂ grown (e) without and (f) with hydrogen, respectively. (g) PL and (h) Raman spectra of ultraclean monolayer MoS₂.

Here we report a two-step CVD process to selectively accomplish lateral and vertical heteroepitaxy between monolayer WS₂ and MoS₂ by careful growth control of monolayer MoS₂ seed crystals. We show that including hydrogen into the carrier gas results in ultraclean MoS₂ monolayers. These ultraclean flakes of MoS₂ suppress the nucleation and growth of additional vertical layers and enable the growth of lateral heteroepitaxial structures of monolayer WS₂ from the MoS₂ edges, creating atomically coherent and sharp in-plane WS₂/MoS₂ heterojunctions. When no hydrogen is used, we obtain MoS₂ monolayers decorated with small particles along the edge area. During the subsequent growth of WS₂, these particles nucleate heteroepitaxial growth of monolayer WS₂ on top of the MoS₂ seed to form vertical WS₂/MoS₂ heterojunctions with perfect alignment.

RESULTS AND DISCUSSION

We synthesize MoS₂ monolayers on *c*-cut sapphire using CVD with MoO₃ and S powder precursors and a carrier gas of Ar or H₂/Ar (4 sccm/20 sccm). Further details of the growth conditions are given in the Supporting Information (SI). In Figure 1, we compare triangular monolayer particles of MoS₂ grown with and without hydrogen in the carrier gas. Under typical CVD conditions without H₂, the edges of the MoS₂ flakes are decorated by small particles, as shown by atomic force microscopy (AFM) in Figure 1a,c. These particles have also been present in previously reported syntheses of MoS₂.^{30–37} Inclusion of 4 sccm of H₂ into the carrier gas during the synthesis leads to the disappearance of these particles, as shown in Figures 1b,d. An additional high-resolution AFM height image of the ultraclean MoS₂ monolayer further confirms the ultraclean surface of the monolayer (Figure S1, SI). In both cases, the flakes are confirmed to be monolayer by the AFM profiles showing a height of approximately 0.7 nm, consistent

with previous reports of monolayer MoS₂.^{30–37} Raman spectroscopy also confirms the monolayer nature of the synthesized materials (Figure 1h). The Raman spectrum of MoS₂ shows two characteristic peaks, the out-of-plane vibration of the sulfur atoms (A₁) and the doubly degenerate in-plane vibrations of the Mo and S atoms (E₂).³⁸ The energy associated with these phonons changes with thickness, and the spectral separation of these two peaks has become a common tool for identifying the number of MoS₂ layers. For our MoS₂ flakes, we see a separation of 20.3 cm⁻¹, in agreement with previously reported syntheses of monolayer MoS₂^{30–37} but slightly larger than exfoliated monolayer samples.³⁹ It is worth noting that the spectrum shown in Figure 1h was taken on resonance with the B exciton absorption band and that resonance Raman spectrum can induce spectral changes including line-broadening, a large fluorescence background, and additional non-zone centered modes. Due to the resonance, we observe additional modes including the mode indicated by “c” in Figure 1h, which are absent in the off-resonance Raman spectrum (Figure S2, SI).^{38,40} To better measure the peak separation, we used an off-resonance 514.5 nm excitation (Figure S2, SI) which removes the spectral congestion and increases signal-to-noise, and we report a spectral separation of 17 cm⁻¹. The source of this decreased splitting is unknown. This could be due to enhanced substrate interactions, but further study is needed to explore this phenomenon. Photoluminescence (PL) spectroscopy further confirms that inclusion of hydrogen leads to a cleaner and more homogeneous material. Figure 1g shows a PL spectrum taken with 632.8 nm laser excitation at room temperature on ultraclean monolayer MoS₂, clearly showing strong A-exciton peak at 667 nm due to the direct band gap of monolayer MoS₂.^{4,7} Parts e and f of Figure 1 show PL intensity maps of the emission from the A exciton (spectrum shown in Figure 1g) of monolayer MoS₂ grown without and with

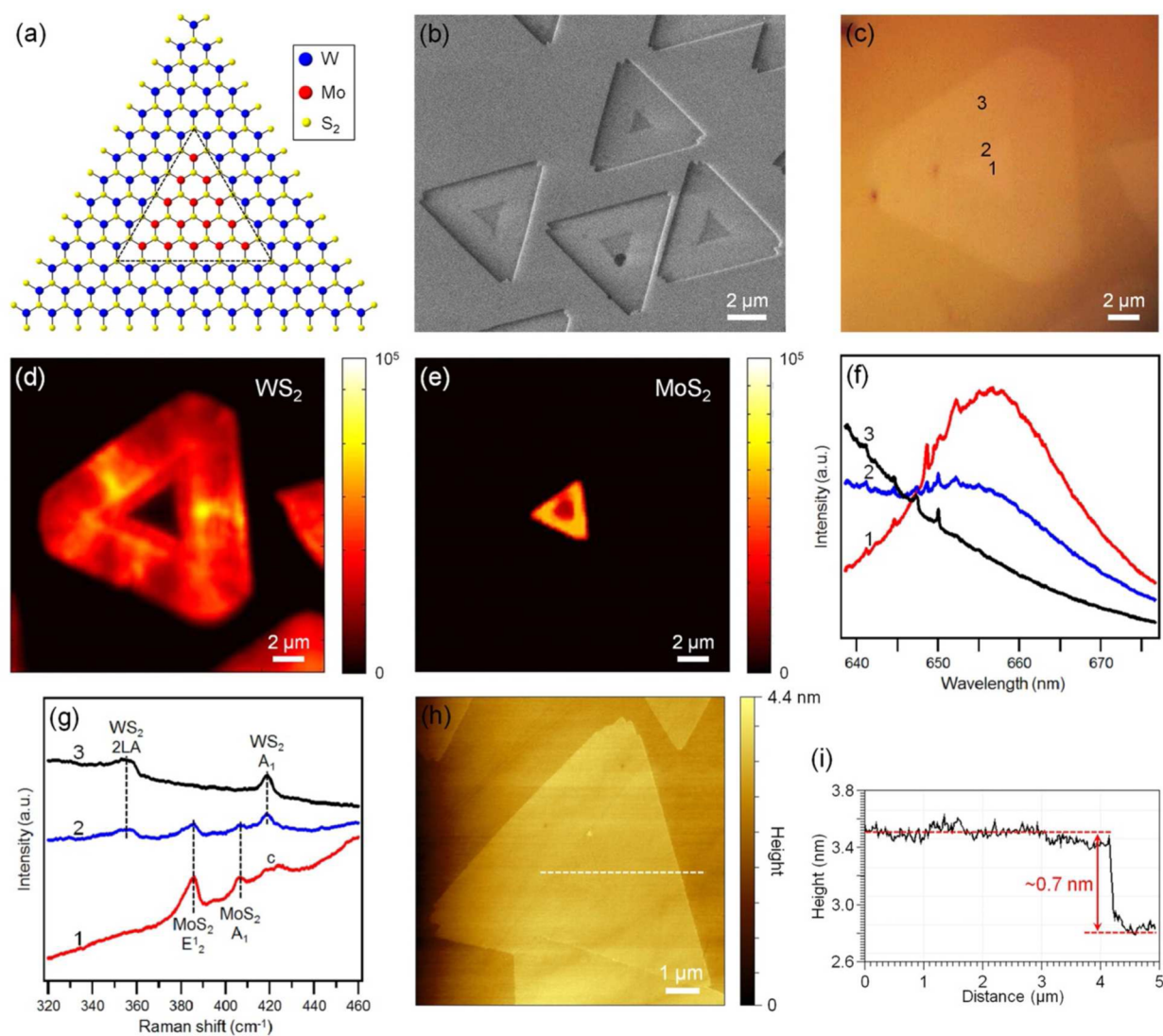


Figure 2. In-plane heteroepitaxial WS_2/MoS_2 monolayers synthesized from monolayer MoS_2 grown with hydrogen. (a) Atomic model of an in-plane heteroepitaxial junction between MoS_2 and WS_2 . (b) SEM image of the in-plane heteroepitaxial monolayers. (c) Optical microscope image of the in-plane heteroepitaxial monolayer. (d and e) PL intensity mapping of WS_2 and MoS_2 from the in-plane heteroepitaxial monolayer, respectively. (f) PL and (g) Raman spectra taken from the points marked by 1–3 in panel c. (h) AFM height image of the in-plane heteroepitaxial monolayer. (i) Height line profile along the dotted white line in panel h.

hydrogen, respectively. In the flakes grown without H_2 , the PL is spatially heterogeneous. Additionally, each flake shows a faint ribbon $\sim 1 \mu\text{m}$ wide with lower intensity due to partial quenching of the PL. When MoS_2 is grown with hydrogen, the PL map is highly uniform across each flake, indicating uniform chemical composition and electronic structure. Scanning electron microscopy (SEM) images of ultraclean and particle-decorated MoS_2 are shown in Figure S3 (SI).

When ultraclean MoS_2 flakes are used as seed particles in the growth of WS_2 , we obtain in-plane, heteroepitaxial WS_2/MoS_2 monolayers, as shown in Figure 2. WS_2 was grown using previously reported low-pressure CVD methods at $1050 \text{ }^\circ\text{C}$ with WO_3 and S powder precursors in a carrier gas of H_2/Ar (5 sccm/60 sccm).⁴¹ Figure 2a shows an atomic model of the in-plane heteroepitaxial WS_2/MoS_2 monolayer obtained by this two-step growth method. The SEM image in Figure 2b shows the in-plane, heteroepitaxial monolayers grown on a *c*-cut sapphire substrate. *c*-cut sapphire has both atomically flat surfaces as well as good lattice matching with MoS_2 and WS_2

and has been shown to improve the crystallinity of CVD-grown WS_2 and MoS_2 .^{42,43} The lattice mismatch is only 0.42 and 0.64% when we consider (3×3) MoS_2 or (3×3) WS_2 supercells on (2×2) sapphire, respectively. The in-plane, heteroepitaxial monolayers are oriented along two preferential directions on the substrate, indicating that the in-plane heteroepitaxial monolayers form epitaxially on *c*-cut sapphire. We frequently observe these oriented in-plane heterostructures on areas of the substrate when we use epitaxial MoS_2 monolayers as seeds.^{42,43} The optical microscope image in Figure 2c shows an in-plane heteroepitaxial monolayer that was used for PL and Raman characterizations. Figure 2f shows PL spectra taken from the points marked by 1–3 in Figure 2c. The PL spectra taken from the inner triangle (point 1) and the outer ribbon (point 3) show strong PL signals of MoS_2 and WS_2 due to emission from the lowest energy excitons (the “A” excitons) monolayer MoS_2 and WS_2 , respectively.^{4,7,39,44} The junction region of the in-plane heteroepitaxial monolayer exhibits PL signals of both WS_2 and MoS_2 . PL spectra taken

under a laser excitation of 514.5 nm are shown in Figure S4 (SI). PL intensity maps of the A exciton from WS₂ and the A exciton from MoS₂ demonstrate the formation of the in-plane WS₂/MoS₂ junction (Figure 2d,e). The PL intensity variation in the maps could be attributed to nonuniform strain that can be induced by the difference of the thermal expansion coefficient between substrates and WS₂/MoS₂ or by the lattice mismatch between substrates and WS₂/MoS₂,^{45,46} which can also lead to the variation of the Raman peak positions (Figure S5, SI). While Raman spectra taken from the inner triangle (point 1 in Figure 2c) shows E₂¹ and A₁ peaks of the MoS₂, Raman spectra from the outer ribbon (point 3 in Figure 2c) show peaks corresponding to the 2LA and A₁ phonons of monolayer WS₂ (Figure 2g). We observe Raman signals of both WS₂ and MoS₂ in the junction region. An AFM height image of the in-plane heterostructure (Figure 2h) and the line profile across the flake (Figure 2i) clearly show that the WS₂/MoS₂ flake is a single monolayer and that both materials are in the same plane with a height of 0.7 nm, the same as the monolayer MoS₂ seed crystals. Surprisingly, the AFM friction image clearly shows the lateral junction between MoS₂ and WS₂ (Figure S6, SI). In addition, Kelvin probe force microscopy (KPFM) data confirm that as-synthesized heterostructures have high-quality lateral junctions (Figure S7, SI).

We used scanning transmission electron microscopy (STEM) to characterize better the interface between these two materials and to demonstrate heteroepitaxy. High-angle annular dark field (HAADF) microscopy can provide contrast between the two materials based on the different scattering cross sections for Mo and W atoms (Z-contrast).⁴⁷ The atomic structure of the in-plane heteroepitaxial monolayer is characterized with atomic resolution by Z-contrast STEM microscopy imaging using an aberration-corrected STEM (FEI Titan, 60 kV). Figure 3a shows a low-magnification HAADF-STEM image of an in-plane heteroepitaxial monolayer. The junction between the outer ribbon of WS₂ and the inner triangle of MoS₂ is visible, but with low contrast. The magnified image of the dotted orange square in Figure 3a clearly shows the contrast between MoS₂ and WS₂ (Figure 3b). The image intensity at each point is determined by the spatially averaged atomic number and the thickness of the sample.⁴⁷ Since the average atomic number of WS₂ is higher than that of MoS₂, the WS₂ has higher image intensity (appears brighter) than the MoS₂ in dark-field imaging. Figure 3c shows an atomic-resolution HAADF-STEM image and the associated fast Fourier-transform (FFT) pattern of the junction region of the in-plane heterostructure, clearly showing the atomically sharp junction between WS₂ and MoS₂ along a zigzag direction. Despite growing the WS₂ at high temperatures, we see only minor annealing and elemental mixing across the interface between WS₂ and MoS₂ with a larger concentration of W substituted into the MoS₂ lattice than Mo substituted into the WS₂ lattice. The FFT patterns show only one set of hexagonal spots, demonstrating that the WS₂ grew from the MoS₂ edges with lattice coherence aided by the small lattice mismatch between WS₂ and MoS₂ (0.22%). Parts d and e of Figure 3 show atomic-resolution HAADF-STEM images and their FFT patterns taken from MoS₂ and WS₂ regions, respectively. Hexagonal lattices of MoS₂ and WS₂ are clearly visible without any substituted atoms. The orientation of the FFT patterns of the WS₂ region is the same as that of the MoS₂ region, further confirming the lattice coherence across the WS₂/MoS₂ boundary.

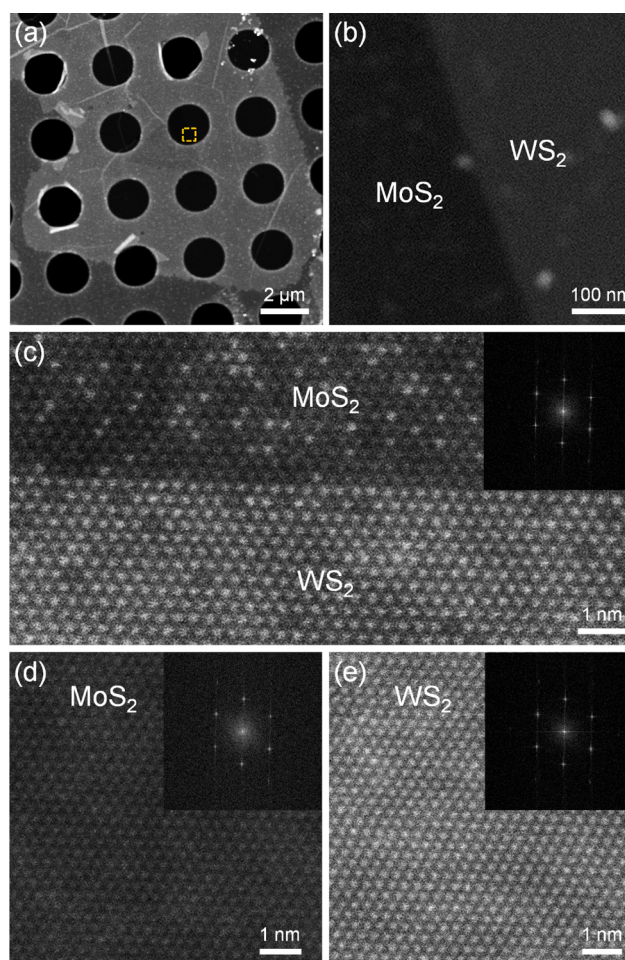


Figure 3. Z-Contrast HAADF-STEM images of the in-plane heteroepitaxial WS₂/MoS₂ monolayer. (a) Low-magnification HAADF-STEM image of the in-plane heteroepitaxial monolayer. (b) Magnified image of the dotted orange square in panel a. (c) Atomic-resolution HAADF-STEM image and its FFT pattern (inset) of the junction region of the in-plane heteroepitaxial monolayer. (d and e) Atomic-resolution HAADF-STEM images and their FFT patterns (inset) of the MoS₂ region and the WS₂ region, respectively, of the in-plane heteroepitaxial monolayer.

When we grow WS₂ using more conventional, particle-decorated MoS₂ monolayers as 2D seed crystals, we obtain vertically stacked WS₂/MoS₂ heterostructures surrounded by a ribbon of monolayer WS₂, as shown in Figure 4. SEM imaging shows a ribbon of WS₂ around the vertical heterostructures on the sapphire substrate (Figure 4a). We identify that these structures are vertically stacked using a combination of PL and Raman spectroscopy and AFM imaging. Figure 4b,c shows PL and Raman spectra taken from different locations of the vertical heterostructure shown in the inset of Figure 4b. The points marked by 1 and 2 indicate the center and edge of the vertical heterostructures, respectively, and each location shows the PL and Raman signals of both WS₂ and MoS₂. The vertical heterostructure shows a very weak PL signal because the PL signal is quenched by charge transfer between the WS₂ and the MoS₂.^{17,48} The small triangular flakes in a neighborhood of the vertical heterostructures (marked as location 3) are identified as WS₂ monolayers by PL and Raman spectroscopy. These particles exhibit a strong PL signal of WS₂ due to a direct band gap of monolayer WS₂, and the Raman spectrum only shows

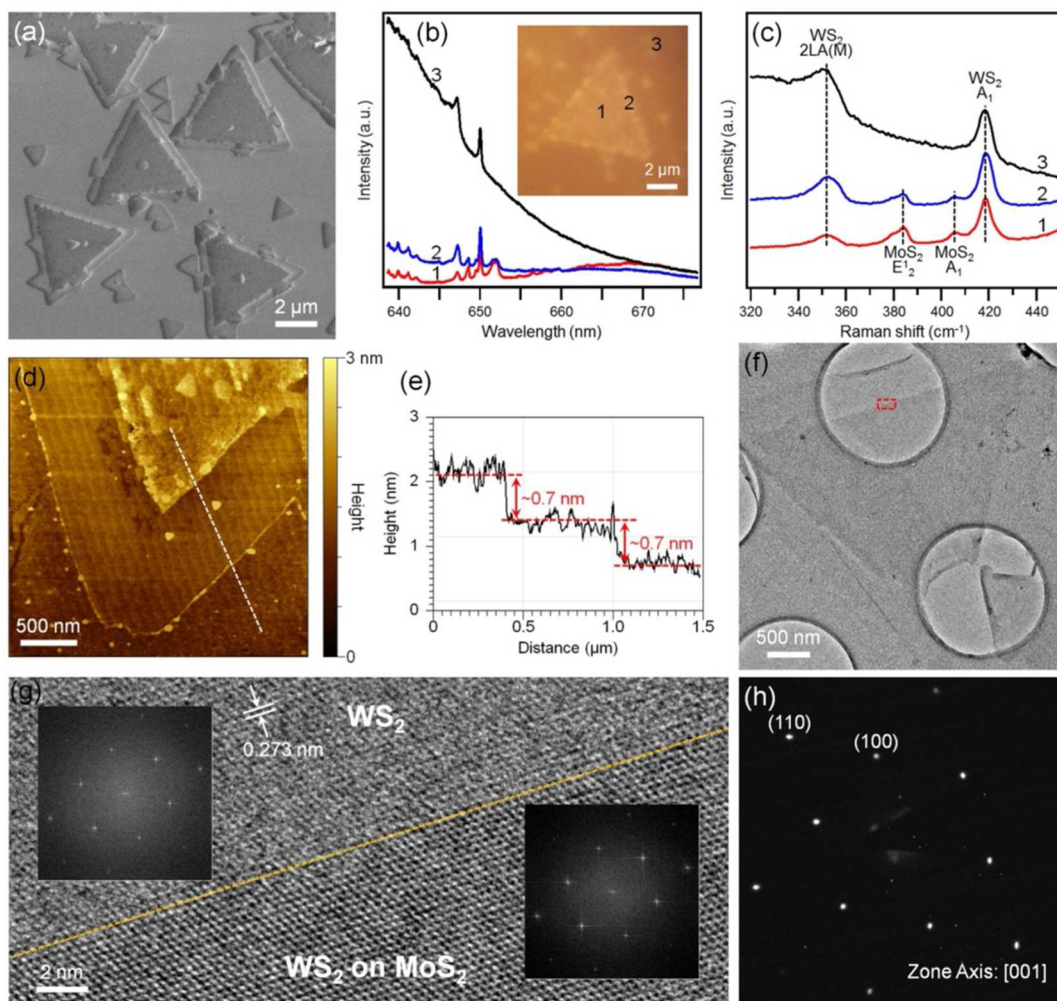


Figure 4. Vertical WS_2/MoS_2 heterostructures synthesized from monolayer MoS_2 grown without hydrogen. (a) SEM image of the vertical heterostructures. (b) PL and (c) Raman spectra taken from the points marked by 1–3 in the inset in panel b. The inset in panel b is an optical microscope image of a vertical heterostructure. (d) AFM height image of a vertical heterostructure. (e) Height line profile along the dotted white line in panel d. (f) Bright-field TEM image of the vertical heterostructure. (g) HRTEM image of the dotted red square in panel f. The insets are FFT patterns of the WS_2 region and the WS_2/MoS_2 region, respectively. (h) SAED pattern of the vertical heterostructure.

peaks from WS_2 and no evidence for MoS_2 . The AFM height image and its line height profile show that the vertical WS_2/MoS_2 heterostructure possesses a thickness of about 1.4 nm and has a monolayer WS_2 ribbon with a thickness of about 0.7 nm (Figure 4d,e).

We used TEM imaging to better characterize the detailed crystal structure of the vertical heterostructures. Figure 4f is a bright-field TEM image of the vertical heterostructure, clearly showing the contrast between an inner triangle of the vertical WS_2/MoS_2 heterostructure and a ribbon of monolayer WS_2 . The contrast of the HAADF-STEM images of the vertical heterostructure is the inverse of the contrast of the HAADF-STEM images of the in-plane heterostructure (Figure S8, SI). Figure 4g is a high-resolution TEM (HRTEM) image of the junction region between the WS_2 ribbon and the WS_2/MoS_2 taken in the region marked by the dotted red square of Figure 4f, showing the hexagonal lattices of the WS_2 and the WS_2/MoS_2 . In the WS_2 region, the lattice spacing of the planes perpendicular to the junction direction is 0.273 nm, which is consistent with the spacing of the (100) planes of WS_2 . The orientation of the FFT patterns of the WS_2 region is the same as that of the WS_2/MoS_2 region, confirming that the ribbon of

monolayer WS_2 forms epitaxially onto the edge of the vertical WS_2/MoS_2 heterostructure. The selected area electron diffraction (SAED) pattern of the vertical WS_2/MoS_2 heterostructure shows only a single set of hexagonal diffraction spots (Figure 4h). This confirms that the WS_2 grows on the top of the MoS_2 with the same stacking orientation, demonstrating vertical heteroepitaxy between WS_2 and MoS_2 .

We believe that in-plane growth of WS_2 ribbons around monolayer MoS_2 is kinetically controlled, because vertical heterostructures are thermodynamically more stable than in-plane heterostructures, as previously reported.²¹ The commonly accepted mechanisms for the formation of MoS_2 and WS_2 monolayers from their oxide precursors are identical. Thermal reduction of the trioxide produces a volatile suboxide.^{49–51} This suboxide can adsorb onto, diffuse along, and desorb from the substrate unless it encounters a nucleation site where it will subsequently sulfurize, forming MoS_2 or WS_2 , respectively. The suboxide clusters can be supplied to the MoS_2 seed flakes by direct impingement from the vapor or by surface diffusion from the substrate. It has been reported that surface diffusion can be the major supply of clusters under high flux conditions while direct impingement can be a major supply of

clusters under low flux conditions.⁵² The suboxide clusters supplied by surface diffusion from the substrate, arriving at the edges of the MoS₂ seed flakes, will contribute mostly to the in-plane growth of WS₂ ribbons around monolayer MoS₂, because the anisotropic flows of materials can induce the anisotropic growth.⁵³ Thus, all the WS₂ growths described above were performed under high flux conditions. By contrast, we obtain vertical WS₂/MoS₂ heterostructures possessing no WS₂ ribbons under low flux conditions (Figure S9, SI). Under our kinetically controlled reaction conditions, monolayer WS₂ grows laterally from the edges of monolayer MoS₂, because these edge sites are the only sites active for nucleation when we use ultraclean MoS₂ monolayers on *c*-cut sapphire as seeds.⁵⁴ This kinetic process is responsible for the formation of the lateral, in-plane heterostructures reported here. When particle-decorated MoS₂ monolayers are used as seeds, not only the MoS₂ edges but also the particles on the MoS₂ surface will serve as nucleation sites, leading to the formation of vertical WS₂/MoS₂ heterostructures with a ribbon of monolayer WS₂.

The ability to control the competitive growth between in-plane and vertical heterostructures hinges on the ability to control the cleanliness of the MoS₂ seed flakes. The small particles that exist on the surface and edges of the monolayer MoS₂ synthesized without hydrogen should be mostly MoS₂ formed by sulfurization of small molybdenum oxide particles. Tiny clusters of molybdenum suboxide will be continuously supplied to the substrate during the reaction. Although most of the suboxide clusters will contribute to the growth of monolayers or desorb from the substrate by re-evaporation due to the high temperature of the substrate, some of the suboxide clusters will aggregate and form more stable small molybdenum oxide particles. During this process, the edge regions of the monolayer MoS₂ could serve as preferential nucleation sites because the edge regions could be sulfur-deficient when no hydrogen is used, as was recently reported.⁵⁵ These particles will subsequently react with sulfur vapor to form small MoS₂ particles. The stability of the suboxide particles is lowered in the presence of highly reducing hydrogen gas.^{49,51} The reduction of suboxide particles by H₂ will revitalize the precursors, leaving the basal plane of the MoS₂ flake clean of debris. In addition, we believe that monolayer MoS₂ synthesized with hydrogen has highly homogeneous edges without sulfur deficiency, because hydrogen can improve the quality of the edges of monolayer TMDCs significantly.^{41,56}

CONCLUSION

In summary, we demonstrate that by carefully controlling contamination and defects of 2D seed crystals we can selectively achieve lateral and vertical heteroepitaxy between monolayer WS₂ and MoS₂ on a *c*-cut sapphire substrate. We show that hydrogen gas plays an important role in removing small particles contaminating MoS₂ monolayer seeds. When we use hydrogen as a carrier gas, we synthesize ultraclean MoS₂ monolayers, which can be used as seeds for lateral heteroepitaxial growth of monolayer WS₂ to form atomically coherent and sharp in-plane WS₂/MoS₂ heterojunctions. When no hydrogen is used, we obtain particle-decorated MoS₂ monolayers, which serve as seeds for vertical heteroepitaxial growth of monolayer WS₂, producing vertical WS₂/MoS₂ heterojunctions. This two-step synthesis can then serve as a building block for making abrupt junctions in 2D materials, as patterned junctions in 2D materials and as platforms for further

exploring the interesting electronic and optical properties of these materials.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/jacs.5b06643.

Synthesis of ultraclean and particle-decorated MoS₂ monolayers; synthetic details of in-plane and vertical WS₂/MoS₂ heterostructures; transfer of WS₂/MoS₂ heterostructures onto a TEM grid; characterization (STEM, TEM, AFM, KPFM, Raman, PL, and SEM measurements); high-resolution AFM height image of monolayer MoS₂ grown with hydrogen; Raman spectra and Lorentz fitting results for monolayer MoS₂; SEM images of monolayer MoS₂ grown with and without hydrogen; PL spectra of in-plane heterostructures taken under a laser excitation of 514.5 nm; Raman peak position mapping of in-plane heterostructures; AFM friction image of the same flake shown in Figure 2g; KPFM data of in-plane heterostructures; vertical heterostructures synthesized under low flux conditions (PDF)

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Notes

The authors declare no competing financial interest.

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